

10.2 reactions of alcohols with hydrogen halides - 10.2 reactions of alcohols with hydrogen halides 441 (10.10a) once the alcohol is protonated, the reaction is an s_n1 reaction with h_2o as the leaving group. when a primary alcohol is the starting material, the reaction occurs as a concerted displacement- **5.2 reactions of alkenes with halogens** - 5.2 reactions of alkenes with halogens 181 this is an electron-pair displacement reaction this example illustrates, the electrophile in this type of reaction both accepts and releases electrons, but the leaving group only accepts electrons. **enzyme kinetics - university science books - home page** - chapter 10 enzyme kinetics one of the most fascinating areas of study in chemical kinetics is enzyme catalysis. the phenomenon of enzyme catalysis usually results in a very large increase in reac-

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